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Background

> Welding fumes

- 574,000 employees in welding, soldering, and brazing occupations in US
- The majority are welders who are exposed to welding fumes.
- Hazardous metals in welding fumes^[1]
- Manganese (Mn): Parkinson's-like disorder
- Iron (Fe): pneumoconiosis
- Zinc (Zn): metal fume fever
- Chromium (Cr): lung cancer (Cr-VI)
- > Metals in toenail as biomarkers
- Inhaled metals deposit in toenails and other body parts.^[2]
- Mn concentration in toenails was proposed as a biomarker of chronic exposure to welding fumes.^[3]

Toenail metal analysis

- No standard methods
- Inductively coupled plasma (ICP)-mass spectrometry (-MS) is widely used.
- ICP-optical emission spectrometry (-OES) or X-ray fluorescence (XRF) can be more appropriate for toenail metal analysis.

Methods	Limit of detection	Cost, time, & labor effectiveness	d
ICP-MS	+++	+	
ICP-OES	++	++	
XRF	+	+++	N

Relevance to National Occupational Research Agenda (NORA)

- Sector programs of "Manufacturing" and "Construction"
- Core and Specialty Program of "Exposure Assessment"
- Cross-sector program of "Cancer, Reproductive, Cardiovascular and Other Chronic Disease Prevention"

Objectives

- > Explore the applicability of ICP-OES and XRF for analyzing metals in toenails in comparison with ICP-MS.
- Hypothesis: ICP-OES and XRF can measure toenail metal concentrations as accurately as ICP-MS.
- Specific Aim 1: Measure metal concentrations in toenails using ICP-OES and compare the results to ICP-MS measurements
- Specific Aim 2: Measure metal concentrations in toenails using XRF and compare the results to ICP-MS measurements

Experimental design

Step 1: Sample collection

- Clip toenails of subjects
- 20 welders and 20 non-welders
- Step 2: Sample cleaning and weighing
- \circ Wash toenails in Triton X–100 (non-ionic surfactant, 1% in deionized water) using ultra-sonication for 1 hour
- Rinse with deionized water (ASTM II) 3 times
- \circ Dry in an oven for > 24 hours and weigh using a microscale





Assessing the Applicability of Methods to Analyze Metals in Toenails

- known concentrations of metals
- Panalytical)

- measurements

- using XRFs and ICPs by toenail mass
- b-XRF against ICP-MS



> Impact of results

- Guideline to select an appropriate method to analyze toenail metals







Ρ	relin	ninary	data								
Comparison of ICP-OES, b-XRF, and p-XRF against ICP-MS											
(welders: n = 6, non-welders: n = 4) Metal concentration in toenails, ug/g											
Method		Mn	Fe	Zn	AI	Cr	Cu	Ni			
ICP-MS		2.6±1.7	36.5±28.0	94.7±27.9	28.2±37.6	1.0±2.6	5.4±1.8	0.1±0.3			
ICP-OES		2.8±2.1	35.3±30.6	81.0±29.3	16.0±32.0	2.2±1.9	6.3±3.0	0.2 ± 0.6			
Pearson's r		1.00	0.96	0.79	0.99	0.22	-0.15	0.19			
b-XRF		2.6±1.4	25.4±6.6	113.2±14.3	0.4±1.9	1.4±0.6	25.8±9.5	0.5 ± 0.5			
Pearson's r		1.00	0.68	0.18	0.3	-0.09	-0.24	-0.10			
р-	XRF	6.6±2.7	32.6±15.8	72.2±35.5	N/A	1.4±0.5	4.9±1.8	0.1 ± 0.0			
Pear	rson's r	0.83	0.93	0.72	N/A	0.51	0.15	0.52			
10 -	Mn •	ICP-OES b-XRF p-XRF	y = 1.13x R ² = 0	+ 3.72 10 .47		-OES RF RF	y = 1.05x - 3 R² = 0.92	.17			
ail, µg	-						ý	= 0.52x + 13.49 R ² = 0.87			
n toen 9	-		y = ' F	1.16x - 0.12 R ² = 0.84	0 +		·····	0.16× 10.46			
ation i			• • • • • • • • • • • • • • • • • • •		0 +	• • • • • • • • • • • • • • • • • • •	y	$R^2 = 0.47$			
centra			y =	$\begin{array}{c} & & \\ 0.75x + 0.63 \\ R^2 = 0.80 \end{array}$	0						
Con	•••••	·····	•	,	••••••	•					
Ŭ	0	1 2	3	4 5		50)	100			
		OES sho	wed strong	correlation v	with ICP-MS	on in toenali m	leasured by ICP	· ⋈ S, µg/g			
	Mn,	Fe, Zn, a	nd Al (r = 1	.00, 0.96, 0.	79, and 0.9	9, respec	ctively).				
(b-XRF showed strong correlation with ICP-MS in 										
	Mn, and Fe (r = 1.00, and 0.68).										
(o p-⊼R Mn,∣	Fe, and Z	Zn (r = 0.83	, 0.93 , and ().72, respec	tively).					
F	uture	e func	ding pla	an							
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		e Tunain stigation	g pian of the relativ	onshin hatw	oon toonail	metal con	acontration	ne and			
C	healt	h outcom	nes (target:	National Ins	titute for Oc	cupation	al Safety a	and			
 Health (NIOSH) R21 or R01 grant) Development of evidence-based policies and regulations for protecting 											
								ing a tool for			
	expo	sure mor	nitoring (tar	get: grant fro	om National	Institutes	s of Health	(NIH),			
	NIH	Health So	ciences (NII	EHS), or Env	vironmental	Protectio	on Agency	(EPA)			
R	efere	ences									
[1] Antonini et al. 2003. Critical Reviews in Toxicology 33 (1): 61-103											
[2] Hopps et al. 1977. Science of the Total Environment 7 (1): 71-89											
[3] \	Nard et	t al. 2018	. Annals of	Work Expos	sures and H	<i>lealth</i> 62	(1): 101–1	1			
	Ackr	nowle	dgeme	nts							
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